

## Note

### A novel method for the determination of the stereochemistry of pyruvate acetal substituents applied to the capsular polysaccharide from *Streptococcus pneumoniae* Type 4

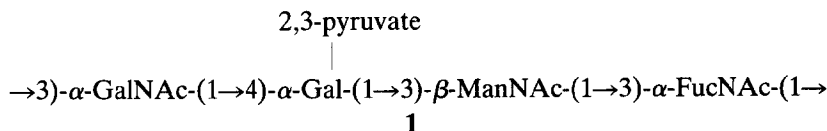
CHRISTOPHER JONES

National Institute for Biological Standards and Control, Potters Bar, Herts., EN6 3QG (Great Britain)

(Received November 3rd, 1988; accepted for publication, August 16th, 1989)

Bacterial polysaccharides often carry non-carbohydrate substituents that affect their chemistry, conformation, and immunology. Pyruvate acetals are amongst the most common substituents and their formation creates a new chiral centre at C-2. Pyruvate acetals have been found 4,6-linked to Gal, Glc, and Man, and 3,4-linked to Gal. The chemical shifts of the  $^{13}\text{C}$  and  $^1\text{H}$  resonances of the methyl group of the pyruvate acetal, or the carboxyl-reduced group, have been correlated with the stereochemistry at C-2, using data from model systems<sup>1,2</sup>.

Less common locations of the pyruvate acetal group occur in the capsular polysaccharides from *Streptococcus pneumoniae* Type 4 (S4, positions 2 and 3 of  $\alpha$ -Gal)<sup>3,4</sup> and *Klebsiella* K1 and K58 (positions 2 and 3 of  $\beta$ -GlcA)<sup>5–7</sup>. The structure of the repeating unit of the S4 polysaccharide was shown to be **1**, but no methods for the determination of the stereochemistry at the pyruvate acetal C-2 were available. A method based on n.O.e. is applicable to these systems and is illustrated by application to the polysaccharide S4.



The  $^1\text{H}$ -n.m.r. spectrum of the deuterium-exchanged polysaccharide was fully assigned from the COSY-45<sup>8</sup>, phase-sensitive NOESY<sup>9</sup>, and 1D n.O.e. spectra at 343 K. The assignments for the  $\alpha$ -Gal and  $\beta$ -ManNAc residues are given in Table I, and Fig. 1 illustrates how they were obtained from the spectra. Of particular note is the coincidence of the  $\alpha$ -Gal H-2 and  $\beta$ -ManNAc H-3 resonances at 4.016 p.p.m. An initial phase-sensitive NOESY experiment, obtained (400 MHz, 343 K) with a mixing time of  $100 \pm 20$  ms, showed extensive spin diffusion (illustrating the rapid proton  $T_2$  relaxation), and was repeated with a mixing time of  $50 \pm 20$  ms. The

rapid relaxation also prevented resolution of the smaller (3–4 Hz) couplings between, for example, the  $\alpha$ -Gal H-1 and H-2. A 1D n.O.e. time-course experiment, with pre-irradiation of the pyruvate methyl for 0.05, 0.10, and 0.15 s, showed a steady build-up in the intensity of the n.O.e. between 0.05 and 0.10 s, but extensive spin diffusion was apparent with a pre-irradiation time of 0.15 s. Fig. 2 shows the 500-MHz 1D  $^1\text{H}$ -n.m.r. spectrum of the polysaccharide together with the n.O.e. difference spectra obtained after pre-irradiation of the  $\alpha$ -Gal H-1 and the pyruvate methyl group. The same Lorentzian-to-Gaussian resolution-enhancement functions were used for the normal and n.O.e. difference spectra. The spectrum with irradiation of the  $\alpha$ -Gal H-1 showed an intense intra-residue n.O.e. and the expected inter-residue n.O.e. across the glycosidic bond to the [ $\alpha$ -Gal H-2 +  $\beta$ -ManNAc H-3] resonance at 4.016 p.p.m.<sup>4</sup>, together with weak enhancements of the  $\beta$ -ManNAc H-1 (4.893 p.p.m.) and H-4 (t, 3.750 p.p.m.) signals, probably due to spin diffusion. The n.O.e. difference spectrum with pre-irradiation of the pyruvate methyl resonance showed an enhancement of the  $\alpha$ -Gal H-3 signal. The small (3–4 Hz)  $\alpha$ -Gal H-1,2,  $\alpha$ -Gal H-3,4, and  $\beta$ -ManNAc H-2,3 couplings are not visible in the normal or n.O.e. difference spectra, and the [ $\alpha$ -Gal H-2 +  $\beta$ -ManNAc H-3] and  $\alpha$ -Gal H-3 resonances appear as doublets with 10–11 Hz ( $J_{2,3}$  for Gal,  $J_{3,4}$  for  $\beta$ -ManNAc). The peaks show the “lean-to” effect expected for second-order effects on tightly coupled systems, as  $\Delta\delta/J_{2,3} = \sim 2.5$  at 500 MHz. Fig. 2 (*d* and *e*) also shows the data obtained from the phase-sensitive NOESY spectrum with a mixing time of 50 ms (400 MHz, 343 K), with a cross-peak between the  $\alpha$ -Gal H-1 and the [ $\alpha$ -Gal H-2 +  $\beta$ -ManNAc H-3] resonance, and a cross-peak between the pyruvate methyl and the  $\alpha$ -Gal H-3. The slices through the NOESY spectrum produced the same data as the 1D n.O.e. difference spectra, but with much poorer resolution. Integration of the n.O.e. difference spectra showed intensities of –26% for the  $\alpha$ -Gal H-1 to [ $\alpha$ -Gal H-2 +  $\beta$ -ManNAc H-3] enhancement, and –8% for the Me to  $\alpha$ -Gal H-3 enhancement, calibrated against the saturation of the irradiated signals. For a macromolecule, these enhancements are negative, with the difference

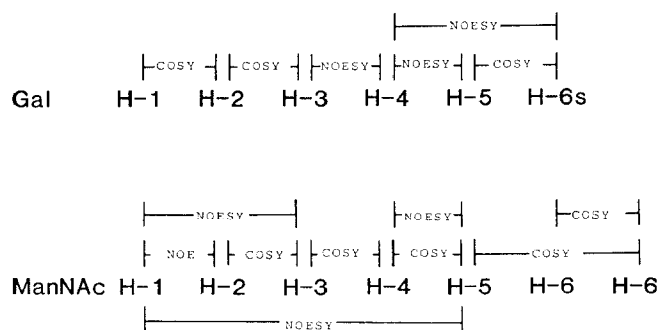


Fig. 1. Basis of the assignment of the resonances of the  $\alpha$ -Gal and  $\beta$ -ManNAc residues. Using COSY-45, and phase-sensitive NOESY (mixing time of 50 ms), 1D n.O.e. difference spectra. Several inter-residue n.O.e.s (e.g.,  $\alpha$ -GalNAc H-1 to  $\alpha$ -Gal H-4) also confirmed the assignments. N.O.e.s used in the determination of the pyruvate acetal stereochemistry are not included.

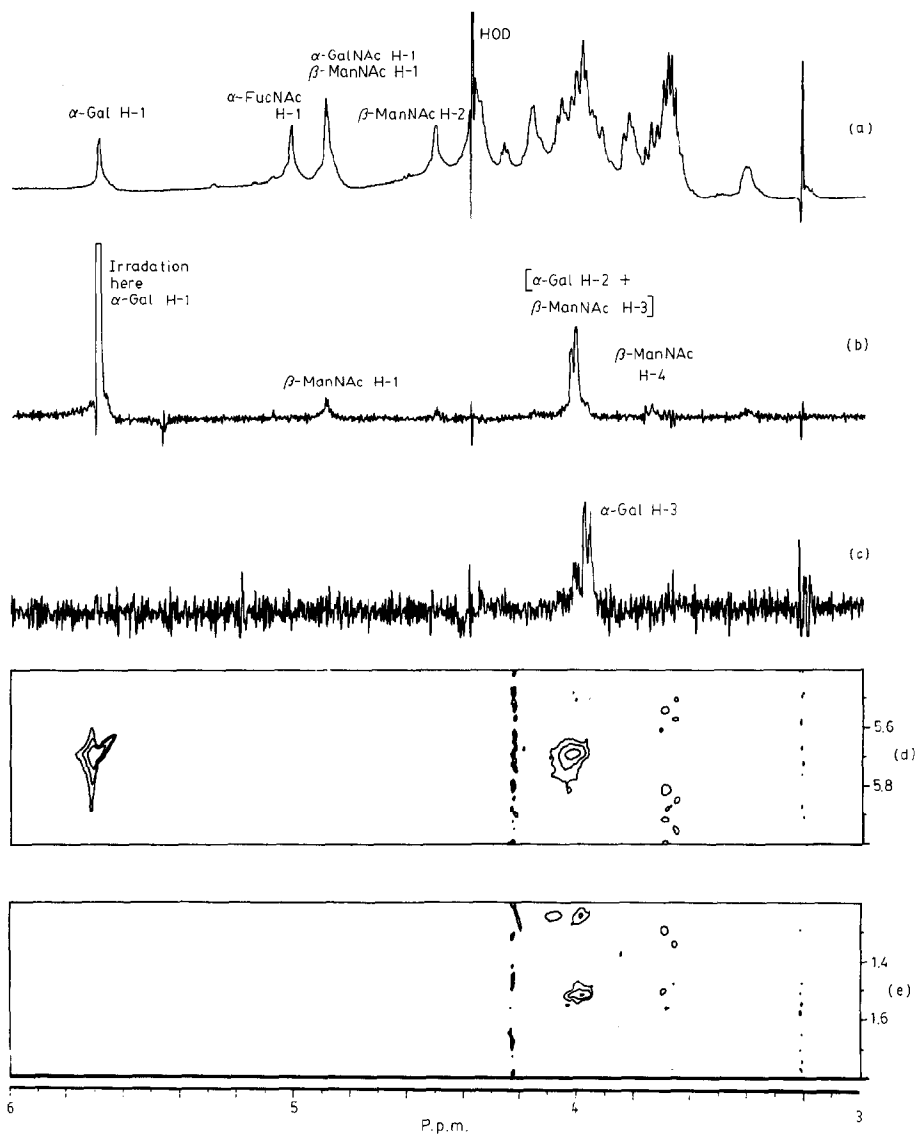


Fig. 2. (a) 500-MHz 1D  $^1\text{H}$ -n.m.r. spectrum of pneumococcal S4 polysaccharide at 343 K, processed with Lorentzian-to-Gaussian resolution enhancement; (b) the 500-MHz n.O.e. difference spectrum obtained with pre-irradiation of the  $\alpha$ -Gal H-1 resonance at 5.693 p.p.m., showing saturation of that resonance and the enhancements of [ $\alpha$ -Gal H-2 +  $\beta$ -ManNAc H-3],  $\beta$ -ManNAc H-1, and  $\beta$ -ManNAc H-4 signals; (c) the 500-MHz n.O.e. difference spectrum obtained with pre-irradiation of the pyruvate methyl resonance at 1.52 p.p.m., showing the enhancement of the  $\alpha$ -Gal H-3 resonance; (d) the partial 400-MHz phase-sensitive NOESY spectrum showing the  $\alpha$ -Gal H-1 resonance on the diagonal and the  $\alpha$ -Gal H-1/([ $\alpha$ -Gal H-2 +  $\beta$ -ManNAc H-3]) cross-peak, and (e) the partial 400-MHz phase-sensitive NOESY spectrum showing the pyruvate methyl/ $\alpha$ -Gal H-3 cross-peak.

TABLE I

PARTIAL PROTON ASSIGNMENT OF THE POLYSACCHARIDE S4 AT 343 K

Residue	H-1	H-2	H-3	H-4	H-5	H-6	H-6'
$\alpha$ -Gal	5.693	4.016	3.959	4.376	4.058	3.695	3.695
$\beta$ -ManNAc	4.893	4.503	4.019	3.750	3.415	3.938	3.823

peaks showing the same phase as the saturated peak in the n.O.e. difference spectra and the cross-peaks the same phase as the diagonal in NOESYPH spectra. No attempt was made to integrate the cross-peak volumes in the NOESY spectra.

The *R* and *S* configurations<sup>10</sup> at pyruvate C-2 are shown in Fig. 3 for  $\alpha$ -Gal in S4. The *S* configuration will lead to enhancement of the  $\alpha$ -Gal H-3, and the observed n.O.e.s are illustrated by arrows. This residue was incorporated into a model of the conformation of the polysaccharide S4<sup>11</sup>, and n.O.e. simulations were carried out using a program which takes into account the free rotation of the methyl group<sup>12</sup>. The n.O.e. at  $\alpha$ -Gal H-3 produced by saturation of the pyruvate methyl group was estimated to be -10%, compared to the observed -8%. The expected ratio of n.O.e.s to  $\alpha$ -Gal H-3 and  $\alpha$ -Gal H-2 was estimated, from simulations, to be 9:1, in good agreement with Fig. 2c. Fig. 3b and n.O.e. simulations suggest that the *R* configuration would result in an enhancement at the  $\alpha$ -Gal H-2 resonance. The local structure in the *Klebsiella* polysaccharides containing pyruvated  $\beta$ -GlcA is very similar and should produce the same pattern of n.O.e.s.

## EXPERIMENTAL

The polysaccharide was obtained from Merck, Sharpe, and Dohme. N.m.r. experiments were performed<sup>4</sup> on a Bruker AM500 or AM400 spectrometer at an indicated\* probe temperature of 343 K. The sample was deuterium-exchanged by lyophilisation of a solution in D<sub>2</sub>O and partially depolymerised in the n.m.r. tube by sonication<sup>13</sup>. 1D N.O.e. difference experiments used a pre-irradiation time of

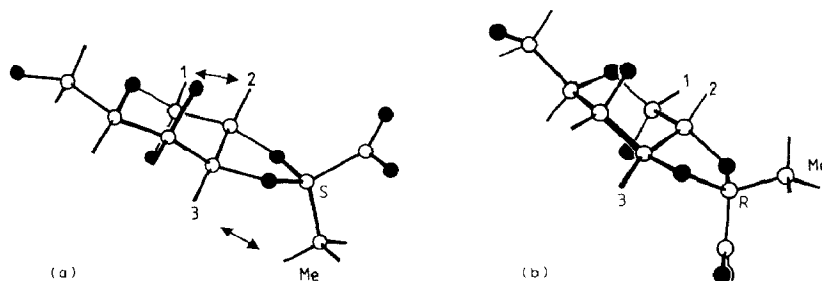


Fig. 3. Computer-generated models for the *S* (a) and *R* (b) 2,3-pyruvate acetals of  $\alpha$ -Gal. The *S* configuration generates an n.O.e. to the  $\alpha$ -Gal H-3 below the ring, whereas the *R* configuration should generate an n.O.e. to  $\alpha$ -Gal H-2.

100 ms and a decoupler power of "35L" for pre-irradiation of the pyruvate methyl and "40L" for pre-irradiation of the  $\alpha$ -Gal H-1. Inter-leaved accumulations of 32 scans up to a total of 320 (pyruvate methyl) or 1280 ( $\alpha$ -Gal H-1) were collected. Quantitation used standard Bruker software and was normalised to the saturated peak. The phase-sensitive NOESY spectrum was obtained using the pulse sequence described by Bodenhausen *et al.*<sup>9</sup>, with 512 time increments, and 32 scans of 2k data points per experiment over a spectral width of 2024 Hz, and zero-filled to a final digital resolution of 2.0 Hz/point in each domain. Mixing times of 100 or 50 ms were used, with a random variation of  $\pm 20$  ms in order to suppress scalar coupling artefacts.

Molecular modelling was carried out on a ChemX system (Chemical Design Ltd., Oxford), run on a MicroVax II, and structures were optimised by the MM2 algorithm<sup>14</sup> (QCPE, Indiana University) optimised for carbohydrates<sup>15</sup>. The modelling was carried out on the free acids. N.O.e. simulations were carried out on a pyruvated deca-saccharide fragment with NOEMOL<sup>12</sup> running on a SUN 3/160. The single correlation time of 7 ns was estimated from integration of intra-residue n.O.e.s and data fitting.

#### ACKNOWLEDGMENTS

Merck, Sharp, and Dohme are thanked for the polysaccharide, the M.R.C. Biomedical NMR Centre, Mill Hill for access to the high-field spectrometers, and Dr. Mark Forster (N.I.B.S.C.) for the n.O.e. simulation software.

#### REFERENCES

- 1 P. J. GAREGG, P.-E. JANSSON, B. LINDBERG, F. LINDH, J. LÖNNGREN, I. KVARNSTROM, AND W. NIMMICH, *Carbohydr. Res.*, 78 (1980) 127-132.
- 2 P. A. J. GORIN, M. MAZUREK, H. S. DUARTE, M. IACOMINI, AND J. H. DUARTE, *Carbohydr. Res.*, 100 (1982) 1-15.
- 3 P.-E. JANSSON, B. LINDBERG, AND U. LINDQUIST, *Carbohydr. Res.*, 95 (1981) 73-80.
- 4 C. JONES AND F. CURRIE, *Carbohydr. Res.*, 184 (1988) 279-284.
- 5 C. ERBING, L. KENNE, B. LINDBERG, J. LÖNNGREN, AND I. W. SUTHERLAND, *Carbohydr. Res.*, 50 (1976) 115-120.
- 6 G. G. S. DUTTON AND A. V. SAVAGE, *Carbohydr. Res.*, 84 (1980) 297-305.
- 7 P.-E. JANSSON, B. LINDBERG, AND G. WIDMALM, *Carbohydr. Res.*, 182 (1988) 166-168.
- 8 W. P. AUE, E. BARTHOLDI, AND R. R. ERNST, *J. Chem. Phys.*, 64 (1976) 2229-2246.
- 9 G. BODENHAUSEN, H. KOGLER, AND R. R. ERNST, *J. Magn. Reson.*, 58 (1984) 370-388.
- 10 R. S. CAHN, C. K. INGOLD, AND V. PRELOG, *Experientia*, 12 (1956) 81-94.
- 11 C. JONES, F. CURRIE, AND M. J. FORSTER, unpublished data.
- 12 M. J. FORSTER, C. JONES, AND B. MULLOY, *J. Mol. Graphics*, 7 (1989) 196-201.
- 13 S. C. SZU, G. ZON, R. SCHNEERSON, AND J. B. ROBBINS, *Carbohydr. Res.*, 152 (1986) 7-20.
- 14 N. L. ALLINGER AND Y. H. YU, *QCPE*, 13 (1981) 395.
- 15 G. A. JEFFREY AND R. TAYLOR, *J. Comput. Chem.*, 1 (1980) 99-109.